



**Physics** is the study of motion, matter, energy, and force.

**Qualitative Descriptions:** are descriptions made by observing with the 5 senses, such as the smell of a flower or the colour of someone's eyes. They include observations which cannot be measured.

**Quantitative Descriptions :** are descriptions that are based on measurements or counting (i.e. they are numerical), such as the number of petals a flower has or how tall a person is. They deal with quantities.

**Le Système International d'Unités (SI)** refers to a single measurement system (metric system) that has been agreed upon by scientist all over the world. SI has seven base units. Most other units are derived from these seven units

**Base unit:** refers to units that are defined.

Table 1. SI base units		
Base quantity	Name	Symbol
SI base unit		
length	meter	m
mass	kilogram	kg
time	second	s
electric current	ampere	A
thermodynamic temperature	kelvin	K
amount of substance	mole	mol
luminous intensity	candela	cd

**Note:**

meter is defined as the distance light travels in a small fraction of a second.

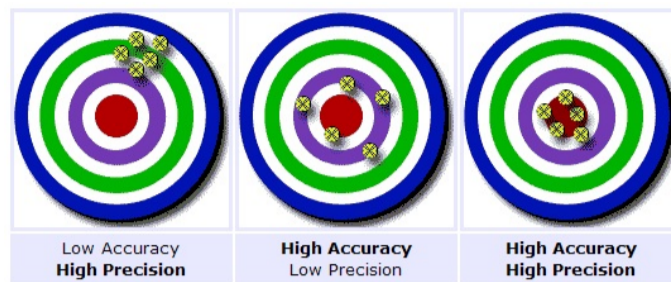
kilogram is the current "base unit" for mass. A kilogram is defined as the mass of a certain lump of platinum and iridium that is kept in Paris under glass to protect it from chemical changes that could alter its mass.

**Derived units** are ones that we "figure out" by using base units.

Derived quantity	Name	Symbol
Table 2. Examples of SI derived units		
SI derived unit		
area	square meter	m <sup>2</sup>
volume	cubic meter	m <sup>3</sup>
speed, velocity	meter per second	m/s
acceleration	meter per second squared	m/s <sup>2</sup>

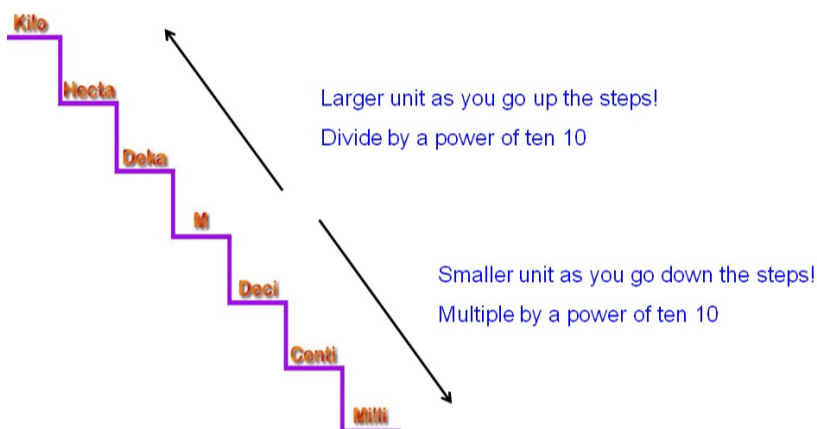
**Accuracy** refers to the closeness of measurements to is how close a measured value is to the actual (true) value.

**Precision** is how close the measured values are to each other.



Converting measurements is a skill that will be tested in high school math and science classes, as well as in some college classes

**Method #1: The Step Stair**



**METHOD #2: CONVERSION FACTOR**

To convert units, we need to multiply the quantity we want to convert by its conversion factor. The conversion factor basically tells us how to convert one unit into another

Example 1:

How many seconds are in seven years?

$$7a \times \frac{365day}{1a} \times \frac{24hours}{1day} \times \frac{60min}{1hr} \times \frac{60}{1min} = 2.2075 \times 10^8 s$$

Example 2:

Convert 30 km/hr to m/s:

$$30 \frac{km}{1hr} \times \frac{1hr}{60min} \times \frac{1min}{60sec} \times \frac{1000m}{1km} = 8.3m / s$$

**General Rule:**

To change from km/hr = m/s ÷ 3.6

To change from m/s to km/hr x 3.6



## PART A: MULTIPLE CHOICE

- Which of the following involves the study of motion, matter, energy, and force?
  - Biology
  - Chemistry
  - Meteorology
  - Physics
- Which of the following is a great physicist?
  - Albert Einstein
  - Galileo Galilei
  - Isaac Newton
  - All are correct
- Which of the following is used to make a qualitative description?
  - Your bath scales
  - Your Eyes
  - A measuring Tape
  - A rain gauge
- Which of the following is a quantitative description?
  - The glass is half full
  - It is warm in the physics lab
  - The lemon tastes sour
  - The mass the cat is 2.0 kg
- Which organization is responsible for creating a system of base units to be followed by the scientific community?
  - International Union of Pure and Applied Chemistry (IUPAC)
  - Le Système International d'Unités (SI)
  - French Academy of Sciences (FAS)
  - International Space Agency (ISA)
- What is the base unit for measuring time?
  - kilograms
  - meter
  - second
  - meter/second
- What is the base unit for measuring mass?
  - kilogram
  - meter
  - second
  - meter/second
- Which of the following is a derived unit?
  - kilograms
  - meter
  - second
  - meter/second

9. Use the picture below to describe accuracy and precision:

	<b>Accuracy</b>	<b>Precision</b>
(A)	Low	Low
(B)	Low	High
(C)	High	Low
(D)	High	High



10. How many seconds are there in 1.5 hours?

- (A) 90 s
- (B) 1500 s
- (C) 5400 s
- (D) 8600 s

11. Convert 1.56 kilograms into grams

- (A) 1560 g
- (B) 156 g
- (C) 1.56 g
- (D) 0.00156 g

12. What is the measurement 455 km, converted to meters?

- (A) 0.000455 m
- (B) 0.455 m
- (C) 45 500 m
- (D) 455 000 m

13. What is 198 km/h equal to?

- (A) 0.0198 m/s
- (B) 55.0 m/s
- (C) 198 m/s
- (D) 7128 m/s

14. What is 120. km/h equal to?

- (A) 0.120 m/s
- (B) 33.3 m/s
- (C) 432 m/s
- (D)  $1.20 \times 10^3$  m/s

1. Complete the chart below. [11]

Step 1. Tell if each of the following is a quantitative or qualitative description.

Step 2. If it is a quantitative description, tell if the unit is a derived unit or a base(standard) unit.

Measurement	Quantitative/ Qualitative	Derived Unit/ Base Unit
a speed of 25 m/s		
a foul odour		
mass is 75.1 kg		
a long trip		
salty taste		
a time of 200.0 seconds		
a density of 200 g/m <sup>3</sup>		

2. Write the correct abbreviation for each metric unit. [9]

- A) Kilogram \_\_\_\_\_ B) Milliliter \_\_\_\_\_ C) Kilometer \_\_\_\_\_  
 D) Meter \_\_\_\_\_ E) Millimeter \_\_\_\_\_ F) Centimeter \_\_\_\_\_  
 G) Gram \_\_\_\_\_ H) Liter \_\_\_\_\_ L) Milligram \_\_\_\_\_

3. Convert the following. [16]

- A) 2000 mg = \_\_\_\_\_ g B) 5 L = \_\_\_\_\_ mL  
 C) 16 cm = \_\_\_\_\_ mm D) 104 km = \_\_\_\_\_ m  
 E) 198 g = \_\_\_\_\_ kg F) 2500 m = \_\_\_\_\_ km  
 G) 480 cm = \_\_\_\_\_ m H) 75 mL = \_\_\_\_\_ L  
 I) 65 g = \_\_\_\_\_ mg J) 5.6 kg = \_\_\_\_\_ g  
 K) 50 cm = \_\_\_\_\_ m L) 6.3 cm = \_\_\_\_\_ mm  
 M) 8.8 mm = \_\_\_\_\_ cm N) 5.6 m = \_\_\_\_\_ cm  
 O) 120 mg = \_\_\_\_\_ g P) 2000 ml = \_\_\_\_\_ L

4. Convert the following

- A) 30.0s = \_\_\_\_\_ min B) 602 min = \_\_\_\_\_ h  
 C) 4.7 h = \_\_\_\_\_ min D) 23.6 h = \_\_\_\_\_ s  
 E) 5024 s = \_\_\_\_\_ min F) 6.2 h = \_\_\_\_\_ min  
 G) 25.40 min = \_\_\_\_\_ h H) 45 km/h = \_\_\_\_\_ m/s  
 I) 2.67 m/s = \_\_\_\_\_ km/h J) 100 km/h = \_\_\_\_\_ m/s  
 K) 15 m/s = \_\_\_\_\_ km/h L) 363 m/s = \_\_\_\_\_ km/h  
 M) 25 km/h = \_\_\_\_\_ m/s N) 2.0 m/s = \_\_\_\_\_ km/h