# SCIENCE 1206 UNIT 3: MOTION WORKSHEET #6: SIGNIFICANT FIGURES AND SCIENTIFIC NOTATION



There are at least two reasons for being familiar with scientific notation.

**36** 

### 1) Method of writing numbers that are very big and very small. It works like this:

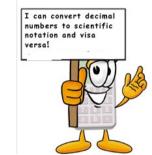
a big number

Speed of light => 
$$300,000,000$$
m/s =  $3.0 \times 10^8$  m

a small number

Charge on an electron  $=> 0.00000000000000001602 \text{ C} = 1.602 \text{ x } 10^{-19} \text{ C}$ 

## 2) helpful for indicating how many significant figures are present in a number



100 cm as  $1.00 \times 10^2$  ( 3 sig fig )cm  $1.0 \times 10^2$  ( 2 sig fig )cm

 $1 \times 10^{2}$  ( 1 sig fig )cm.

### PART A: MULTIPLE CHOICE

[12]

- 1. Which of the following is equal to  $3.26 \times 10^4 \text{ m}$ ?
  - (A) 0.000326 m
  - (B) 0.00326 m
  - (C) 32600 m
  - (D) 326000 m
- 2. Which of the following is equal to  $7.01 \times 10^{-3} \text{ C}$ ?
  - (A) 0.00701 C
  - (B) 0.0701 C
  - (C) 7.01 C
  - (D) 701 C
- 3. Which of the following is not correctly expressed in scientific notation?
  - (A)  $27.01 \times 10^3 \text{ s}$
  - (B)  $4.2 \times 10^9 \text{ s}$
  - (C)  $3.09 \times 10^{-2} \text{ s}$
  - (D)  $6.1 \times 10^4 \text{ s}$
- 4. Which of the following is the largest number?
  - (A)  $3.9 \times 10^6 \text{ kg}$
  - (B)  $3.99 \times 10^{-7} \text{ kg}$
  - (C)  $3.1 \times 10^{10} \text{ kg}$
  - (D)  $4 \times 10^6 \text{ kg}$
- 5. Which of the following is the smallest number?
  - (A)  $1.1 \times 10^6$
  - (B)  $9.6 \times 10^{-10}$
  - (C)  $4.2 \times 10^3$
  - (D)  $4.9 \times 10^{-6}$

- 6. Mr. Smith construction company was contracted to build a building that ended up having a mass o 7400 kg. How would you express this in scientific notation?
  - (A)  $7.4 \times 10^2 \text{ kg}$
  - (B)  $74 \times 10^2 \text{ kg}$
  - (C)  $7.4 \times 10^3 \text{ kg}$
  - (D)  $740 \times 10 \text{ kg}$
- 7. A satellite is measured at 205,000 km away from the Earth. How would you express this in scientific notation?
  - (A)  $205 \times 10^3$
  - (B)  $20.5 \times 10^4$
  - (C)  $2.05 \times 10^5$
  - (D) 2.05 x 104
- 8. How would you express 6.28 x 10<sup>-4</sup> m in standard form?
  - (A) -0.000628 m
  - (B) -62800 m
  - (C) 62800 m
  - (D) 0.000628 m
- 9. How many significant digits is in  $5.98 \times 10^{24} \text{ kg}$ ?
  - (A) 3
  - (B) 4
  - (C) 5
  - (D) 6
- 10. How would you express 213.49 °C in scientific notation:
  - (A)  $2.1349 \times 10^{2} \, ^{\circ}\text{C}$
  - (B)  $0.21349 \times 1^{3} \, ^{\circ}\text{C}$
  - (C)  $2.13 \times 10^{2} \, {}^{\circ}\text{C}$
  - (D)  $2.1349 \times 10^{-2} \, ^{\circ}\text{C}$
- 11. What is the measurement 101 000 grams in scientific notation?
  - (A)  $1.01 \times 10^5 \text{ g}$
  - (B)  $1.0100 \times 10^{-5} g$
  - (C)  $1.01000 \times 10^5 \text{ g}$
  - (D)  $10.1 \times 10^4 \text{ g}$
- 12. Solve:  $123\ 000\ \text{m}\ \text{x}\ 3\ 234\ \text{m} = ?$ 
  - (A)  $39800000 \text{ m}^2$
  - (B)  $3.98 \times 10^8 \text{ m}^2$
  - (C)  $3.97 \times 10^{-7} \text{ m}^2$
  - (D)  $398 \text{ m}^2$

## PART B: WRITTEN RESPONSE

1.	Convert each of the following into scientific notation:		
	a)	300 000 000 m/s	
	b)	0.00000000000000000016 C	
	c)	47 045 mm	
	d)	4.05 m	
	e)	25 m	
	f)	0.0305 kg	
	g)	0.0082 s	
	h)	243N	
2.	Write the following in standard form:		[8]
	a)	2.8 x 10 <sup>3</sup> m/s	
	b)	2.130 x 10 <sup>-2</sup> m	
	c)	3.7 x 10 <sup>2</sup> mm	
	d)	5.05 x10 <sup>-3</sup> m	
	e)	2.15 x 10 <sup>4</sup> m	
	f)	3.34 x10 <sup>0</sup> kg	
	g)	8.2 x 10 <sup>-3</sup> s	
	h)	2.33 10 <sup>1</sup> s N	
3.	Express the answer to each of the following calculations with the correct number of significant figures in scientific notaion.		
	a)	3.0 cm x 4.000 cm	[8]
	b)	0.0045mm x 0.90 mm	
	C)	2.005 cm x 5.0 cm	
	d)	1.452 m ÷ 8.2 s	
	e)	0.465 m ÷ 0.03000 s	
	f)	92.2 kg x 293.00 m/s	
	g)	2.73 J ÷ 458 C	
	h)	18.00 N x 351 s	